

## Rietveld refinement of Ba<sub>5</sub>(AsO<sub>4</sub>)<sub>3</sub>Cl from high-resolution synchrotron data

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Key indicators: powder synchrotron study;  $T = 298$  K; mean  $\sigma(\text{As}-\text{O}) = 0.040$  Å;  $R$  factor = 0.059;  $wR$  factor = 0.082; data-to-parameter ratio = 22.1.

The apatite-type compound Ba<sub>5</sub>(AsO<sub>4</sub>)<sub>3</sub>Cl, pentabarium tris(arsenate(V)) chloride, has been synthesized by ion exchange at high temperature from a synthetic sample of mimetite (Pb<sub>5</sub>(AsO<sub>4</sub>)<sub>3</sub>Cl) with BaCO<sub>3</sub> as a by-product. The results of the Rietveld refinement, based on high resolution synchrotron X-ray powder diffraction data, show that the title compound crystallizes in the same structure as other halogenoapatites with general formula A<sub>5</sub>(YO<sub>4</sub>)<sub>3</sub>X (A = divalent cation, Y = pentavalent cation, X = Cl, Br) in space group *P6<sub>3</sub>/m*. The structure consists of isolated tetrahedral AsO<sub>4</sub><sup>3-</sup> anions (*m* symmetry), separated by two crystallographically independent Ba<sup>2+</sup> cations that are located on mirror planes and threefold rotation axes, respectively. The Cl<sup>-</sup> anions are at the *2b* sites ( $\bar{3}$  symmetry) and are located in the channels of the structure.

### Related literature

For crystal chemistry of apatites, see: Mercier *et al.* (2005); White & ZhiLi (2003); Wu *et al.* (2003). For powder diffraction data on Ba-containing As-apatites, see: Kreidler & Hummel (1970); Dunn & Rouse (1978). Atomic coordinates as starting parameters for the Rietveld (Rietveld, 1969) refinement of the present phases were taken from Chengjun *et al.* (2005); Dai *et al.* (1991); de Villiers *et al.* (1971). For related Ba–Cl-apatites, see: Đorđević *et al.* (2008); Hata *et al.* (1979); Reinen *et al.* (1986); Roh & Hong (2005); Schiff-Francois *et al.* (1979). For synthetic work, see: Baker (1966); Essington (1988); Harrison *et al.* (2002).

### Experimental

#### Crystal data

As<sub>3</sub>Ba<sub>5</sub>ClO<sub>12</sub>  
 $M_r = 1138.85$   
Hexagonal, *P6<sub>3</sub>/m*  
 $a = 10.5570$  (1) Å  
 $c = 7.73912$  (8) Å  
 $V = 746.98$  (1) Å<sup>3</sup>  
 $Z = 2$   
Synchrotron radiation

$\lambda = 0.998043$  Å  
 $\mu = 56.07$  (1) mm<sup>-1</sup>  
 $T = 298$  K  
Specimen shape: cylinder  
40 × 0.7 × 0.7 mm  
Specimen prepared at 100 kPa  
Specimen prepared at 1258 K  
Particle morphology: powder, white

#### Data collection

In-house design diffractometer  
Specimen mounting: capillary  
Specimen mounted in transmission mode

Scan method: step  
Absorption correction: none  
 $2\theta_{\min} = 2$ ,  $2\theta_{\max} = 70^\circ$   
Increment in  $2\theta = 0.01^\circ$

#### Refinement

$R_p = 0.059$   
 $R_{wp} = 0.082$   
 $R_{exp} = 0.067$   
 $R_B = 0.090$   
 $S = 1.23$   
Excluded region(s): 2-6 degrees  $2\theta$ .

Profile function: Fundamental  
Parameters  
464 Bragg reflections  
21 parameters  
Preferred orientation correction: none

**Table 1**

Selected geometric parameters (Å, °).

Ba1–O1	2.67 (5)	Ba2–O1 <sup>v</sup>	3.14 (4)
Ba1–O2 <sup>i</sup>	2.81 (4)	Ba2–Cl1 <sup>iv</sup>	3.281 (5)
Ba1–O3 <sup>i</sup>	3.12 (3)	As1–O3	1.64 (2)
Ba2–O2 <sup>ii</sup>	2.59 (4)	As1–O1	1.70 (8)
Ba2–O3 <sup>iii</sup>	2.62 (4)	As1–O2	1.70 (4)
Ba2–O3 <sup>iv</sup>	3.05 (4)		
O3–As1–O3 <sup>vi</sup>	118 (2)	O3–As1–O2	108 (2)
O3–As1–O1	108 (1)	O1–As1–O2	106 (2)

Symmetry codes: (i)  $x - y, x, -z$ ; (ii)  $-y + 1, x - y + 1, z$ ; (iii)  $y, -x + y + 1, -z$ ; (iv)  $x, y + 1, z$ ; (v)  $-x + y, -x + 1, z$ ; (vi)  $x, y, -z + \frac{1}{2}$ .

Data collection: local software; cell refinement: *CELREF* (Laugier & Bochu, 2003); data reduction: local software; method used to solve structure: coordinates taken from a related compound; program(s) used to refine structure: *TOPAS* (Coelho, 2000); molecular graphics: *Balls and Sticks* (Kang & Ozawa, 2003); software used to prepare material for publication: *publCIF* (Westrip, 2008).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WM2188).

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